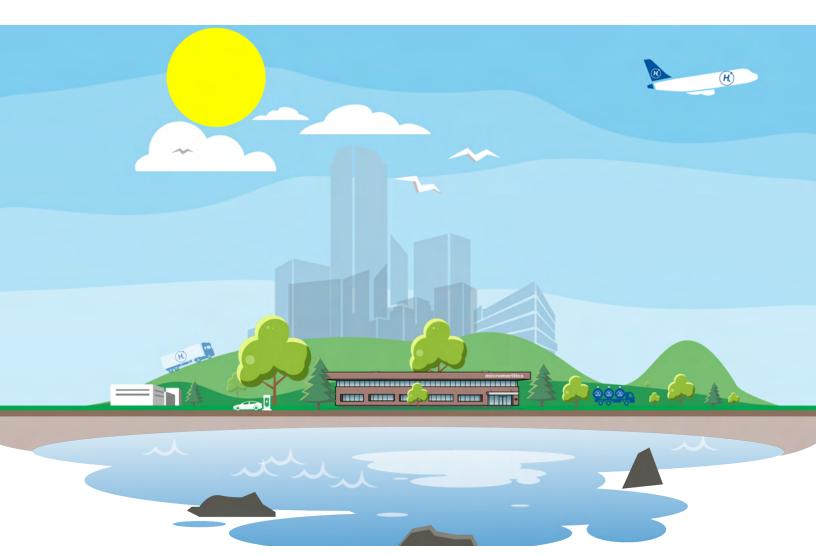
NET-ZERØ TECHNOLOGIES

Micromeritics offers the <u>most comprehensive</u> portfolio of <u>high-performance instruments</u> to characterize the materials required to achieve a more <u>sustainable future</u>





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HYDROGEN LIFE CYCLE



Hydrogen will play a key role in decarbonization as it supports 60% of the applications with greenhouse gas (GHG) emissions.

Micromeritics products play a key role in the development of Adsorbents, Membranes, and **Catalysts** critical for technology development.

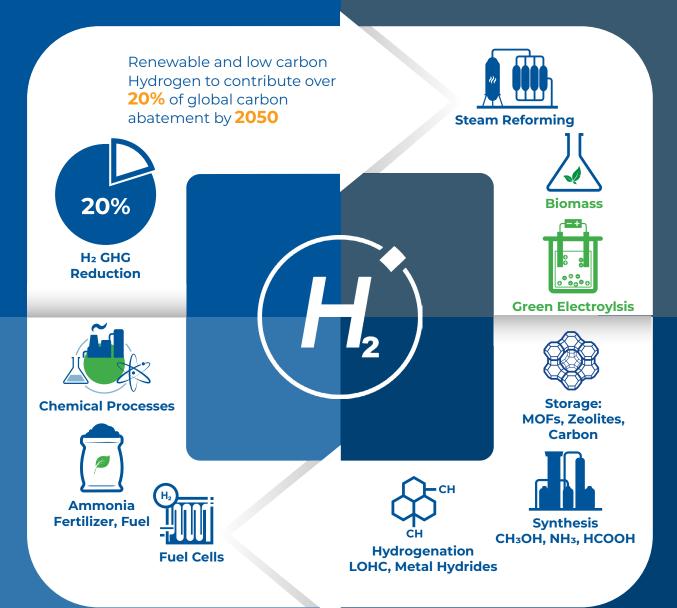
Adsorbents, Membranes, and Catalysts

- Optimize pore size of fuel cell membranes
- Use chemisorption to determine catalyst active area
- Adsorb/Desorb cycle optimization to minimize costs
- Study fuel cell efficiencies

HYDROGEN APPLICATION



HYDROGEN PRODUCTION



HYDROGEN STORAGE





Blue Hydrogen is derived from natural gas with CO₂ capture and Green Hydrogen is produced by water electrolysis using renewable electricity.

Adsorbents, Membranes, and Catalysts

- Optimize adsorption / desorption cycle to increase productivity and reduce cost
- Determine CO₂ that can be adsorbed
- Maximize activity and lifetime of the catalyst
- Measure membrane pore size to optimize transport and reactivity

Adsorbents, Catalysts

- Develop materials with high H₂ adsorption
- Determine critical parameters to scale adsorbents
- Understand efficiency and lifetime of catalysts
- Maximize catalytic activity



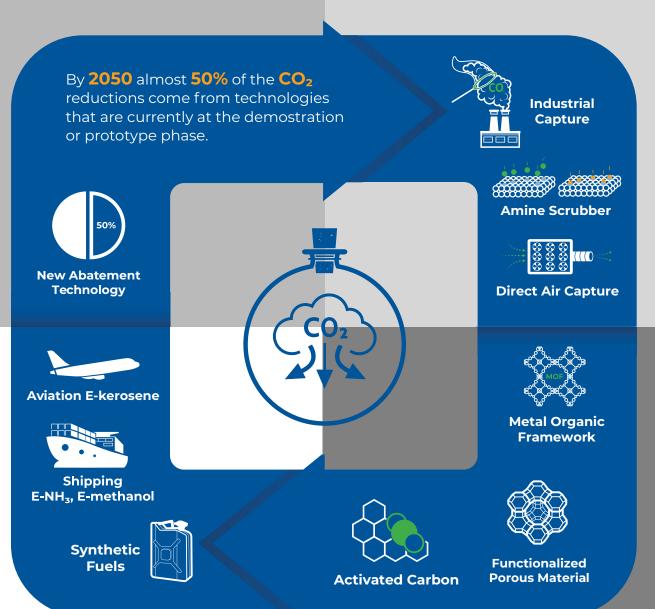
CARBON DIOXIDE MITIGATION







Carbon capture, utilization, and storage, CCUS, is an important portfolio of emissions reduction technologies. A clean energy future includes electric vehicles, **valorizing CO₂** for synthethic fuels, and industrial plants using carbon capture.



Catalysts

- Evaluate effects of temperature, pressure, and time on stream on process economics
- Textural characterization of catalyst support
- Ascertain deactivation mechanisms
- Optimize metal dispersion and activity
- Determine reaction kinetics, activity, and selectivity of the catalyst



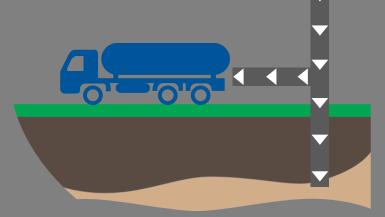


 CO_2 STORAGE

Adsorbents. Membranes

- Effect of water on performance
- Tailor pore size of membrane for application
- Optimize adsorption / desorption cycle to minimize cost

- Determine lifetime, cycling performance and adsorbent CO₂ capacity
- Understand local pollutants effect on adsorbent cycle life



CO₂

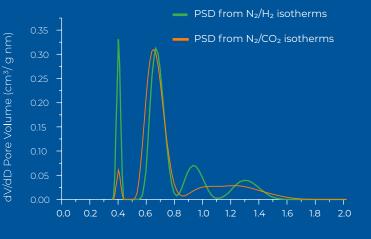
3FLEX

High-performance adsorption analyzer for measuring surface area, pore size and volume

- Understand adsorbent process cost using isoteric heat of adsorption
- Optimize pore size to maximize uptake capacity of the adsorbent
- Predict the selectivity of a gas mixture using Ideal Adsorption Solution Theory (IAST)



COMPLETE PORE SIZE DISTRIBUTION (PSD) USING DUAL NLDFT FOR ACTIVATED CARBON



Pore Diameter (nm)

ADSORBENT AND MEMBRANE SOLUTIONS

BreakThrough Analyzer (BTA)

Precise characterization of adsorbents or membranes under process relevant conditions

- Lifetime and cycling studies to choose best adsorbent technology
- Measure kinetic performanceof adsorbents
- Understand humidity effects for CO₂/N₂ competitive adsorption

6

60 60 60

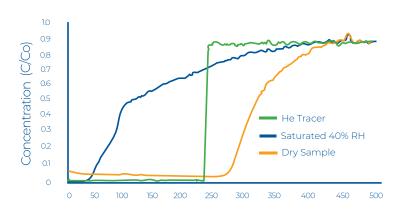
AutoPore

Mercury porosimetry analysis permits detailed porous material characterization

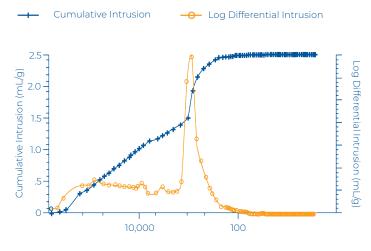
- Characterize pore size to understand diffusion into adsorption sights
- Study and optimize pore size distribution, total pore volume, percent porosity, particle size, and total surface area
- Assure reproducible adsorbent
 manufacturing process



CO₂ BREAKTHROUGH CURVES SIAI LOADED WITH PEI



NaY ZEOLITE CUMULATIVE INTRUSION VS PORE SIZE



Time (minutes)

Pore Size Diameter (nm)

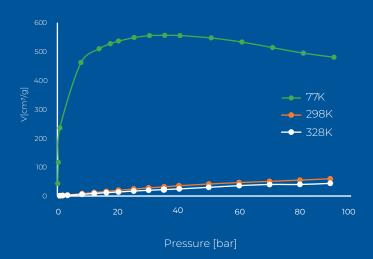


Static volumetric method to obtain high pressure adsorption and desorption isotherms

- Investigate the quantity of H_2 or CO_2 adsorbed
- Increase productivity and reduce cost by optimizing the adsorption / desorption cycle
- Study candidate materials and CO₂ storage sites



H₂ ADSORPTION ON MICROPOROUS CARBON



* Not all products and configurations are available in all regions

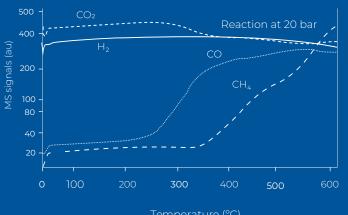
FR/MR REACTOR SYSTEMS

Benchtop reactor studies to understand and optimize catalyst performance

- Understand reaction kinetics to optimize operating parameters and conversion
- Measure selectivity, efficiency, and lifetime of catalysts
- Study reactions requiring gas / liquid separation at temperature and pressure



REDUCTION OF CO₂ IN THE SABATIER REACTION



Temperature (°C)

CATALYST SOLUTIONS

ICCS

Provides in-situ characterization to understand the effect of reaction conditions on the catalyst

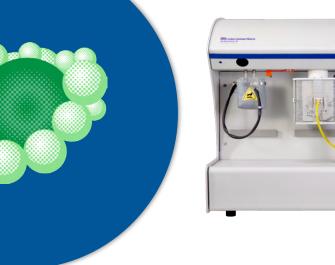
- Understand changes in performance over extended periods
- Determine deactivation mechanism to maximize the catalysts' lifetime
- Monitor changes in active sites, oxidative state, metal dispersion, and desorption behavior

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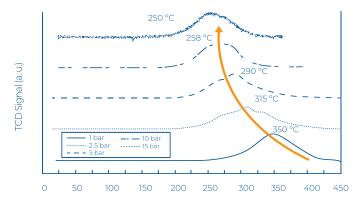
AutoChem

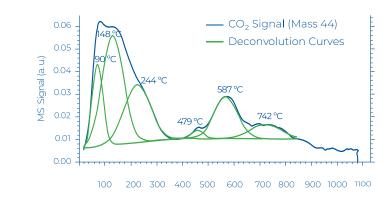
Utilizes dynamic techniques to characterize materials' active sites

- Optimize adsorption and dissociation of H_2/O_2 on electrolysis electrodes
- Understand if desorption occurs near reaction conditions
- Measure and quantify acid or base sites to optimize reactivity and selectivity



DECONVOLUTION OF CO₂ DESORBED BY CaO/MgO





Temperature (°C)

Temperature (°C)

PRESSURE IMPACT ON REDUCTION TEMPERATURE **Cu-OXIDE CATALYST**

3Flex CHEMISORPTION

Offers physisorption and static/dynamic chemisorption for characterizing catalysts

- Understand multi-metal catalysts' effects on activation and adsorption of active species
- Select catalysts providing a higher turnover frequency
- Investigate influence of heat of adsorption



ANALYSIS OF A SUPPORTED NI CATALYST USING H₂



Pressure (mmHq)

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